Cross-Dehydrogenative Coupling of Azoles with α -C(sp³)–H of Ethers and Thioethers under Metal-Free Conditions: Functionalization of H– N Azoles via C–H Activation

Hariprasad Aruri,^{†,‡} Umed Singh,^{†,‡} Sumit Sharma,^{†,‡} Satish Gudup,[†] Mukesh Bhogal,[†] Sanjay Kumar,[†] Deepika Singh,[§] Vivek K. Gupta,^{||} Rajni Kant,^{||} Ram A. Vishwakarma,^{†,‡} and Parvinder Pal Singh^{*,†,‡}

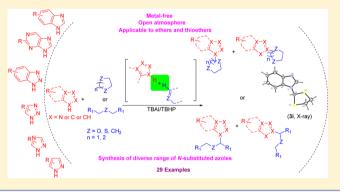
[†]Medicinal Chemistry Division, CSIR-Indian Institute of Integrative Medicine, Canal Road, Jammu-18001, India

[‡]Academy of Scientific and Innovative Research, Canal Road, Jammu-18001, India

[§]Quality Control and Quality Assurance, CSIR-Indian Institute of Integrative Medicine, Canal Road, Jammu-18001, India

Supporting Information

ABSTRACT: A metal-free cross-dehydrogenative coupling method for the synthesis of *N*-substituted azoles has been developed. The TBAI/TBHP system catalyzed the coupling of azoles with ethers and thioethers via α -C(sp³)–H activation. Under the optimized conditions, a diverse range of un/ substituted azoles such as 1*H*-benzimidazole, 9*H*-purine, 1*H*-benzotriazole, 1*H*-1,2,3-triazole, 1*H*-1,2,4-triazole, and 1*H*-pyrazole were successfully employed for coupling with various ethers and thioethers such as tetrahydrofuran, tetrahydropyran, 1,4-dioxane, diethyl ether, tetrahydrothiophene, and 1,3-dithiolane.



N-Substituted azoles, particularly N-alkylated azoles, represent an important class of compounds because of their common occurrence in medicinally important products (some examples are given in Figure 1)¹ as well as their use as precursors for Nheterocyclic carbenes and ionic liquids.² In view the importance of N-substituted azoles, the development of new synthetic methods for the functionalization of azoles is of great interest. Traditionally, N-alkylated azoles have been synthesized by coupling of azoles with electrophiles (Figure 2). These reactions normally require harsh conditions and often result in over-alkylation.³ In the past decade, there has been great interest in the exploration and development of metal-catalyzed and metal-free organocatalyzed C-H bond activation/ functionalization methods for C–C, C–O, and C–N bonds.^{4–6} In this direction, Pan et al.⁷ applied an Fe/TBHPcatalyzed C-H activation strategy for the N-alkylation of azoles (as shown in Figure 2). Moreover, in recent years, metal-free organocatalytic systems such as tetrabutylammonium iodide (TBAI)/tert-butyl hydroperoxide (TBHP) have also been extensively explored for C-H activation/functionalization.⁸

In continuation of our interest in the development of C–H activation/functionalization methods,⁹ here we have developed a general cross-dehydrogenative coupling (CDC) method for the coupling of azoles with α -C(sp³)–H of ethers and thioethers under metal-free conditions using TBAI in the presence of TBHP to active the α -C(sp³)–H of ethers and thioethers. The present method has wide applicability and

successfully works with diverse un/substituted azoles such as 1*H*-benzimidazole, 9*H*-purine, 1*H*-benzotriazole, 1*H*-1,2,3-triazole, 1*H*-1,2,4-triazole, and 1*H*-pyrazole.

To optimize the reaction conditions, 1H-benzimidazole (1) and tetrahydrofuran (THF, 2) were used as coupling partners, and the results of the optimization studies are summarized in Table 1. In the first instance, coupling of 1 with THF in the presence of 10 mol % TBAI and 2 equiv of 70% aqueous TBHP in 1,2-dichloroethane (DCE) gave the expected coupled product 3a in 45% yield (Table 1, entry a), as confirmed by NMR and MS analyses. Increasing the amount of TBHP from 2 to 3.5 equiv afforded a 72% yield of the coupled product 3a (entry b). Further increases in the amount of TBHP did not affect the formation of 3a (results not shown). Changing the amount of THF also affected the yield, as decreasing the amount of THF from 15 to 5 equiv lowered the yield of 3a from 72% to 60% (entries c and d). When the reaction was performed in neat THF, no improvement was noticed (entry e). Changing the solvent from DCE to ethyl acetate (EtOAc) also did not show any improvement (entry f). The effect of temperature was also studied: when the reaction was conducted at rt, the coupled product 3a was observed in 50% yield, but the reaction took a comparatively longer time (entry g). Changing the source of iodide ions to sodium iodide (NaI) or potassium

Received: November 12, 2014 Published: January 14, 2015

The Journal of Organic Chemistry

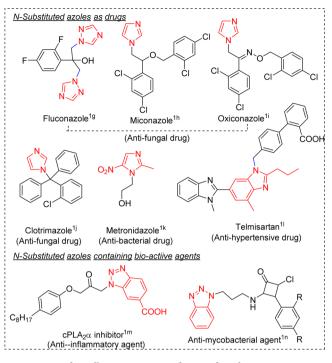


Figure 1. Medicinally important *N*-substituted azoles.

iodide (KI) was also tried, but neither gave better results (entries h and i). With molecular iodine (I_2), the expected coupled product **3a** was observed in only 55% yield (entry j). In the absence of TBHP, no coupling was observed (entry k). On the basis of the above results, the best coupling conditions involved 15 equiv of THF, 10 mol % TBAI, and 3.5 equiv of TBHP in DCE as the solvent at 80 °C.

To know the generality of the optimized conditions, the couplings of various un/substituted 1*H*-benzimidazoles with ethers and thioethers were also investigated, and the results are given in Table 2. The presence of substitution on the 1*H*-benzimidazole greatly affects the formation of the coupled products. Coupling of 5,6-dimethyl-1*H*-benzimidazole with

Table 1. Optimization Studies^a

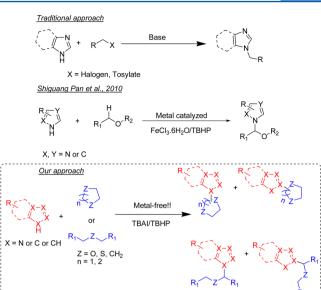


Figure 2. Previous and present approaches for the alkylation of *N*-azoles.

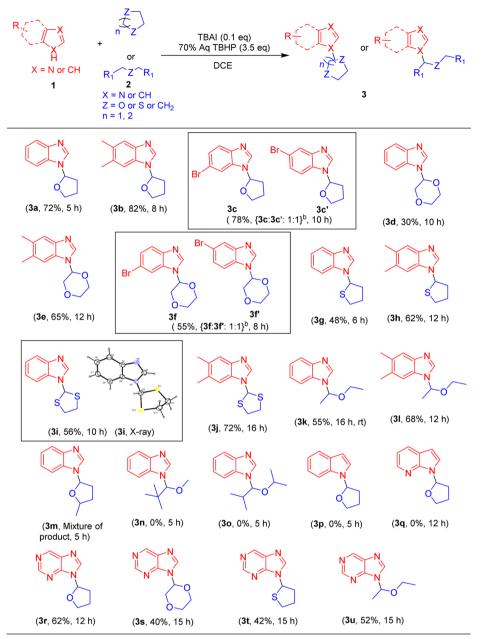
THF gave the corresponding product 3b in an excellent yield of 82%. Coupling of 6-bromo-1*H*-benzimidazole with THF gave a mixture of unseparable regioisomers 3c and 3c' in an overall yield of 78%. 1*H*-Benzimidazole 1 underwent coupling with 1,4-dioxane to give 3d in 30% yield, while on the other hand, coupling of 5,6-dimethyl-1*H*-benzimidazole with 1,4dioxane afforded 3e in 65% yield. Similarly, coupling of 6bromo-1*H*-benzimidazole to 1,4-dioxane gave an unseparable mixture of regioisomers 3f and 3f' in an overall yield of 55%.

Under the optimized conditions, un/substituted 1Hbenzimidazoles also underwent coupling with tetrahydrothiophene to give the corresponding coupled products in moderate to good yields. When 1H-benzimidazole **1** was coupled with tetrahydrothiophene, the corresponding product **3g** was observed in 48% yield. 5,6-Dimethylbenzimidazole underwent coupling with tetrahydrothiophene to give **3h** in 62% yield. Benzimidazole **1** also underwent coupling with 1,3-dithiolane to

| | | N H H | Catalyst/oxidant Solvent, 6 -8 h | | | |
|----------------|-----------------|-------------|-------------------------------------|---------|--------|------------------------|
| | | 1 | 2 | 3a | | |
| entry | ether 2 (equiv) | catalyst | oxidant (equiv) ^c | solvent | T (°C) | yield (%) ^d |
| a | THF (15) | TBAI (0.1) | TBHP (2) | DCE | 80 | 45 |
| b | THF (15) | TBAI (0.1) | TBHP (3.5) | DCE | 80 | 72 |
| с | THF (10) | TBAI (0.1) | TBHP (3.5) | DCE | 80 | 70 |
| d | THF (5) | TBAI (0.1) | TBHP (3.5) | DCE | 80 | 60 |
| e ^b | THF | TBAI (0.1) | TBHP (3.5) | - | 80 | 70 |
| f | THF (15) | TBAI (0.1) | TBHP (3.5) | EtOAc | 80 | 70 |
| g | THF (15) | TBAI (0.1) | TBHP (3.5) | DCE | rt | 50 |
| h | THF (15) | NaI (1) | TBHP (3.5) | DCE | 80 | 65 |
| i | THF (15) | KI(1) | TBHP (3.5) | DCE | 80 | 65 |
| j | THF (15) | $I_{2}(1)$ | TBHP (3.5) | DCE | rt | 55 |
| k | THF (15) | TBAI (0.1) | - | DCE | 80 | 0 |

^aAll of the reactions were performed with 1 mmol of azole. ^bPerformed using THF as the solvent. ^c70% aqueous TBHP. ^dIsolated yields.

Table 2. Coupling of Un/substituted Azoles with Ethers/Thioethers^a



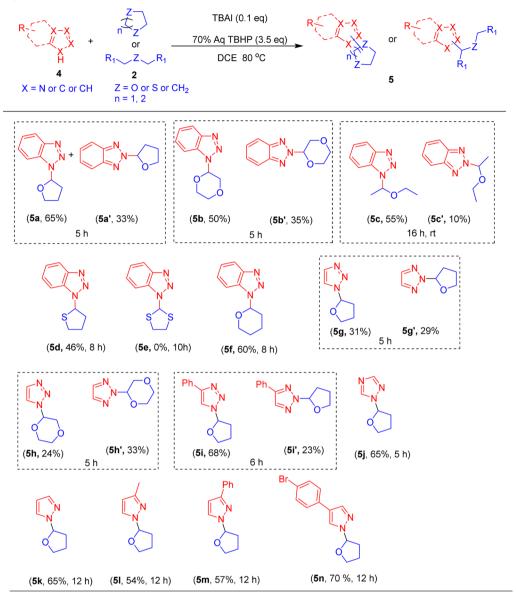
^aReaction conditions (unless otherwise noted): azole 1 (1 mmol), ether/thioether 2 (15 mmol), DCE (15 mL), 80 °C. Isolated yields are shown. ^bThe ratio was revealed by ${}^{1}H/{}^{13}C$ NMR spectroscopy.

afford 1-(1,3-dithiolan-2-yl)-1*H*-benzo[*d*]imidazole (**3i**) in 56% yield. The structure of **3i** was also confirmed by X-ray crystallography (details are given in the Supporting Information). 5,6-Dimethylbenzimidazole also underwent coupling with 1,3-dithiolane to give **3j** in 72% yield. The optimized conditions also worked with alicyclic ethers such as diethyl ether, giving corresponding coupled product **3k** in 55% yield. Coupled product **3l** was formed in 68% yield by the reaction between 5,6-dimethylbenzimidazole and diethyl ether. When a substituted cyclic ether such as 2-methyltetrahydrofuran was tried, a mixture of products was observed. Apart from diethyl ether, other alicyclic ethers such as diisopropyl ether and *tert*-butyl methyl ether did not work under the optimized conditions. *N*-Heterocycles such as indole and 7-azaindole also did not work under the optimized conditions. Interestingly

when purine was tried, the coupling reaction proceeded, giving the coupled products in good to moderate yields. Coupling of purine with THF, 1,4-dioxane, tetrahydrothiophene, and diethyl ether gave the corresponding coupled products **3r**, **3s**, **3t**, and **3u** in yields of 62%, 40%, 42% and 52%, respectively. 1*H*-Benzimidazoles containing electron-donating groups gave comparatively better yields.

Next, we extended the reaction to other azoles, such as 1H-1,2,3-benzotriazole and various 1H-1,2,3-triazoles, and the optimized method worked efficiently, giving separable regioisomeric mixtures of the corresponding coupled products in moderate to good yields (Table 3). When 1H-1,2,3-benzotriazole was treated with THF under the optimized conditions, a separable mixture of regioisomers **5a** and **5a**' was observed in 65% and 33% yield, respectively. 1,4-Dioxane also

Table 3. Coupling of Un/substituted Azoles with Ethers/Thioethers⁴



^aReaction conditions (unless otherwise noted): azole 4 (1 mmol), ether/thioether 2 (15 mmol), DCE (15 mL), 80 °C. Isolated yields are shown.

underwent coupling smoothly to give a separable mixture of regioisomers 5b and 5b' in 50% and 35% yield, respectively. Similarly, 1H-1,2,3-benzotriazole also underwent coupling with diethyl ether to give a mixture of regioisomers 5c and 5c' in 55% and 10% yield, respectively. 1H-1,2,3-Benzotriazole also underwent coupling with tetrahydrothiophene to afford the coupled product 5d in 46% yield as the major isolable regioisomer. 1,3-Dithiolane did not work with benzotriazole, but pyran coupled successfully to give 5f in 60% yield as the major isolable regioisomer. 1H-1,2,3-Triazole underwent coupling with THF to give the corresponding separable mixture of regioisomers 5g and 5g' in 31 and 29% yield, respectively. Under the optimized conditions, 1H-1,2,3-triazole also coupled with 1,4-dioxane, and a mixture of regioisomers 5h and 5h' was isolated in 24 and 33% yield, respectively. 4-Phenyl-1H-1,2,3-triazole also underwent coupling with THF, giving a separable mixture of regioisomers 5i and 5i' in 68 and 23% yield, respectively. To our delight, the optimized method also worked with 1H-1,2,4-triazole, which underwent coupling

with THF to give the corresponding coupled product **5j** in 65% yield. Coupling of 1*H*-pyrazole with THF under the optimized conditions gave the corresponding coupled product **5k** in 65% yield. 3-Methyl- and 3-phenyl-1*H*-pyrazole also underwent coupling with THF to give the respective coupled products **51** and **5m** in 54 and 57% yield, respectively. 4-(4-Bromophenyl)-1*H*-pyrazole also worked under the optimized conditions, giving coupled product **5n** in 70% yield.

To gain insight into the reaction mechanism involved for the coupling of azoles with ethers/thioethers, some experiments were conducted. When the coupling of benzimidazole with THF was performed in the presence of the free-radical scavenger 2,2,6,6-tetramethylpiperidin-1-oxyl (TEMPO), the yield of the coupled product **3a** decreased drastically and the formation of the THF–TEMPO coupled product **6** was observed, confirming the involvement of a free-radical pathway. Surprisingly, even in the presence of 10 equiv of TEMPO, the formation of the coupled product was not completely suppressed (Scheme 1), indicating the initial formation of

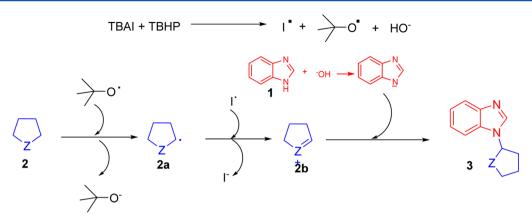
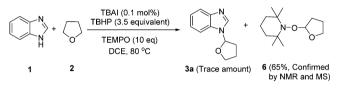


Figure 3. Plausible mechanism for the coupling of azoles with ethers/thioethers.

Scheme 1. Coupling in the Presence of a Free-Radical Scavenger



radical **2a** followed by immediate conversion into cation **2b** (as observed earlier),^{7,8} which is attacked by the azole to give the *N*-alkylated product. On the basis of the literature^{7,8} and our experimental findings, a plausible mechanism would involve the intermediacy of a radical/oxonium or thionium ion and follow the path depicted in Figure 3.

In conclusion, we have developed a general, metal-free, TBAI/TBHP-catalyzed method for the *N*-alkylation of azoles via α -C(sp³)–H activation of ethers and thioethers. The optimized method works successfully with diverse ranges of un/substituted azoles and ethers/thioethers. Moreover, the present method gives a diverse range of *N*-substituted azoles that can be explored for pharmacological applications and as synthons/intermediates in organic synthesis.

EXPERIMENTAL SECTION

General Information. ¹H, ¹³C, and DEPT NMR spectra were recorded on a 400 MHz spectrometer using CDCl₃ as the solvent. Melting points were recorded on a digital melting point apparatus. HRMS spectra were recorded on a UHD-Q-TOF HRMS instrument. MALDI MS spectra were recorded on MALDI-TOF mass spectrometer using 2,5-dihydroxybenzoic acid/ α -cyano-4-hydroxycinnamic acid as the matrix in acetonitrile/water containing 0.01% TFA. GC–MS data were obtained using a GC–EI mass spectrometer. Three-dimensional X-ray intensity diffraction data for a block-shaped single crystal were recorded on a X-ray diffractometer equipped with a CCD camera.

General Procedure for the Synthesis of Compounds in Table 2. Azole 1 (1*H*-benzimidazole, 5,6-dimethyl-1*H*-benzimidazole, 6-bromo-1*H*-benzimidazole, or 9*H*-purine) (1 mmol), ether/thiother 2 (15 mmol), TBHP (3.5 mmol, 70% in water), and TBAI (0.01 mmol) were dissolved in 8 mL of DCE solvent. The reaction mixture was refluxed until the starting material was consumed. After completion of the reaction, the solvent was evaporated from the reaction mixture, and the residue was extracted with ethyl acetate solvent. The product was purified by flash chromatography using a 1% methanol and DCM solvent system.

General Procedure for the Synthesis of Compounds in Table 3. Azole 4 (1*H*-benzotriazole, 1*H*-1,2,3-triazole, 1*H*-1,2,4-triazole, pyrazole) (1 mmol), ether/thioether **2** (15 mmol), TBHP (3.5 mmol, 70% in water), and TBAI (0.01 mmol) were dissolved in 8 mL of DCE solvent. The reaction mixture was refluxed until the starting material was consumed. After completion of the reaction, the solvent was evaporated from the reaction mixture, and the residue was extracted with ethyl acetate solvent. The product was purified by flash chromatography using a 15% ethyl acetate and hexane solvent system. (Note: we observed that products containing 1H-1,2,3-triazole underwent degradation upon storage at room temperature).

Spectral Data. 1-(Tetrahydrofuran-2-yl)-1H-benzo[d]imidazole (3a) (IIIM/724/1572/CN/18).¹ Colorless liquid; TLC $R_f = 0.5$ (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 8.03 (s, 1H), 7.85–7.75 (m, 1H), 7.50–7.42 (m, 1H), 7.36–7.25 (m, 2H), 6.19 (dd, J = 5.6, 4.0 Hz, 1H), 4.23–4.17 (m, 1H), 4.10–4.04 (m, 1H), 2.52–2.44 (m, 2H), 2.19–2.12 (m, 2H); ¹³C NMR (100.61 MHz, CDCl₃) δ 144.1, 140.2, 132.6, 123.1, 122.5, 120.3, 110.5, 86.0, 68.9, 31.8, 24.2; HRMS (ESI+) calcd for $C_{11}H_{13}N_2O$ 189.1028 ([M + H]⁺), found 189.1028.

5,6-Dimethyl-1-(tetrahydrofuran-2-yl)-1H-benzo[d]imidazole (**3b**) (IIIM/724/1572/CN/65). Colorless liquid; TLC $R_{\rm f}$ = 0.6 (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 7.90 (s, 1H), 7.55 (s, 1H), 7.23 (s, 1H), 6.14 (t, *J* = 4.7 Hz, 1H), 4.17 (dd, *J* = 14.7, 7.1 Hz, 1H), 4.05 (dd, *J* = 15.3, 7.7 Hz, 1H), 2.50–2.30 (m, 8H), 2.15 (dd, *J* = 14.3, 7.1 Hz, 2H); ¹³C NMR (100.61 MHz, CDCl₃) δ 142.9, 139.5, 132.2, 131.3, 131.2, 120.3, 110.6, 85.9, 68.8, 31.7, 24.2, 20.5, 20.2; HRMS (ESI+) calcd for C₁₃H₁₇N₂O 217.1341 ([M + H]⁺), found 217.1349.

6-Bromo-1-(tetrahydrofuran-2-yl)-1H-benzo[d]imidazole (**3c**) (*IIIM*/724/1572/CN/48) and 5-Bromo-1-(tetrahydrofuran-2-yl)-1H-benzo[d]imidazole (**3c**') (*IIIM*/724/1572/CN/48). The ratio was calculated by ¹H and ¹³C NMR analysis. Colorless liquid; TLC R_f = 0.5 (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 7.99 (s, 1H), 7.91 (s, 0.47H), 7.63 (d, *J* = 8.7 Hz, 1H), 7.33 (dd, *J* = 22.5, 8.5 Hz, 1.51H), 6.10 (dd, *J* = 5.8, 2.8 Hz, 1H), 4.19–4.09 (m, 1H), 4.08–3.99 (m, 1H), 2.50–2.30 (m, 2H), 2.19–1.98 (m, 2H); ¹³C NMR (100.61 MHz, CDCl₃) δ 141.2, 140.8, 126.1, 125.8, 123.0, 121.4, 116.4, 115.5, 113.7, 111.8, 86.1, 69.0, 31.8, 24.1; HRMS (ESI+) calcd for C₁₁H₁₂BrN₂O 267.0133 ([M + H]⁺), found 267.0139.

1-(1,4-Dioxan-2-yl)-1H-benzo[d]imidazole (**3d**) (IIIM/724/1572/ CN/27).¹⁰ Colorless liquid; TLC $R_{\rm f}$ = 0.5 (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 8.21 (s, 1H), 7.75 (d, *J* = 6.0 Hz, 1H), 7.44 (d, *J* = 5.6 Hz, 1H), 7.29–7.18 (m, 2H), 5.63 (s, 1H), 4.07 (d, *J* = 4.6 Hz, 2H), 3.88–3.59 (m, 4H); ¹³C NMR (100.61 MHz, CDCl₃) δ 143.2, 141.4, 133.1, 123.6, 122.9, 120.3, 110.6, 78.5, 67.9, 66.4, 63.4; HRMS (ESI+) calcd for C₁₁H₁₃N₂O₂ 205.0977 ([M + H]⁺), found 205.0978.

1-(1,4-Dioxan-2-yl)-5,6-dimethyl-1H-benzo[d]imidazole (**3e**) (*IIIM*/724/1572/CN/67). Colorless liquid; TLC $R_f = 0.5$ (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 8.16 (s, 1H), 7.57 (s, 1H), 7.28 (s, 1H), 5.65 (t, *J* = 3.6 Hz, 1H), 4.45 (t, *J* = 6.1 Hz, 1H), 4.24–4.06 (m, 1H), 3.97–3.75 (m, 4H), 2.38 (d, *J* = 7.3 Hz, 6H); ¹³C NMR (100.61 MHz, CDCl₃) δ 142.5, 140.8, 132.7, 131.7, 131.4,

The Journal of Organic Chemistry

120.3, 110.7, 78.5, 68.1, 66.4, 63.4, 20.5, 20.2; HRMS (ESI+) calcd for $C_{13}H_{17}N_2O_2$ 233.1290 ([M + H]⁺), found 233.1293.

6-Bromo-1-(1,4-dioxan-2-yl)-1H-benzo[d]imidazole (**3f**) (IIIM/ 724/1572/CN/61) and 5-Bromo-1-(1,4-dioxan-2-yl)-1H-benzo[d]imidazole (**3f**) (IIIM/724/1572/CN/61). The ratio was calculated by ¹H and ¹³C NMR analysis. Colorless liquid; TLC $R_f = 0.5$ (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ ¹H NMR (400 MHz, CDCl₃) δ 8.19 (d, J = 8.8 Hz, 1H), 7.90 (s, 0.46H), 7.66–7.57 (m, 1H), 7.35 (d, J = 4.8 Hz, 1.30H), 5.66–5.55 (m, 1H), 4.15–3.99 (m, 2H), 3.89–3.63 (m, 4H); ¹³C NMR (100.61 MHz, CDCl₃) δ 142.6, 142.2, 126.5, 126.2, 123.3, 121.7, 116.8, 115.8, 113.9, 111.9, 78.5, 67.8, 66.4, 63.1; HRMS (ESI+) calcd for C₁₁H₁₂BrN₂O₂ 283.0082 ([M + H]⁺), found 283.0084.

1-(*Tetrahydrothiophen-2-yl*)-1*H-benzo*[*d*]*imidazole* (**3***g*) (*IIIM*/ 724/1572/CN/23). Colorless liquid; TLC $R_f = 0.5$ (5% MeOH/ DCM); ¹H NMR (400 MHz, CDCl₃) δ 8.25 (s, 1H), 7.74 (d, *J* = 7.4 Hz, 1H), 7.40 (d, *J* = 6.9 Hz, 1H), 7.28–7.09 (m, 2H), 6.00 (s, 1H), 3.29–3.16 (m, 1H), 2.97 (dd, *J* = 16.0, 9.3 Hz, 1H), 2.39–2.15 (m, 4H); ¹³C NMR (100.61 MHz, CDCl₃) δ 144.3, 141.8, 133.3, 122.9, 122.5, 120.5, 110.0, 63.0, 38.2, 33.0, 28.7; HRMS (ESI+) calcd for C₁₁H₁₃N₂S 205.0799 ([M + H]⁺), found 205.0797.

5,6-Dimethyl-1-(tetrahydrothiophen-2-yl)-1H-benzo[d]imidazole (**3h**) (IIIM/724/1572/CN/68). Colorless liquid; TLC $R_f = 0.5$ (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 8.09 (s, 1H), 7.48 (s, 1H), 7.14 (s, 1H), 5.99–5.87 (m, 1H), 3.22 (dd, *J* = 10.2, 6.2 Hz, 1H), 2.96 (dd, *J* = 16.0, 9.2 Hz, 1H), 2.41–2.23 (m, 8H), 2.22–2.08 (m, 1H), 2.06–1.89 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 142.1, 140.0, 131.0, 130.8, 130.3, 119.5, 109.1, 61.9, 37.1, 31.9, 27.6, 19.6, 19.2; HRMS (ESI+) calcd for $C_{13}H_{17}N_2S$ 233.1112 ([M + H]⁺), found 233.1128.

1-(1,3-Dithiolan-2-yl)-1H-benzo[d]imidazole (**3i**) (IIIM/724/1572/ CN/38). White solid; mp 110 °C; TLC $R_f = 0.5$ (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 8.27 (s, 1H), 7.73 (dd, J = 6.6, 2.0 Hz, 1H), 7.42 (dd, J = 6.6, 1.9 Hz, 1H), 7.29–7.22 (m, 2H), 7.00 (s, 1H), 3.54–3.32 (m, 4H); ¹³C NMR (100.61 MHz, CDCl₃) δ 144.5, 141.4, 132.8, 123.3, 122.9, 120.6, 110.4, 65.8, 39.5; HRMS (ESI+) calcd for C₁₀H₁₁N₂S₂ 223.0364 ([M + H]⁺), found 223.0368.

1-(1,3-Dithiolan-2-yl)-5,6-dimethyl-1H-benzo[d]imidazole (**3***j*) (*IIIM*/724/1572/CN/69). White solid; mp 135 °C; TLC $R_{\rm f}$ = 0.5 (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 8.22 (s, 1H), 7.54 (s, 1H), 7.24 (s, 1H), 7.01 (s, 1H), 3.58–3.40 (m, 4H), 2.39 (d, *J* = 16.5 Hz, 6H); ¹³C NMR (100.61 MHz, CDCl₃) δ 143.1, 140.7, 132.5, 131.8, 131.6, 120.6, 110.5, 65.8, 39.5, 20.6, 20.2; HRMS (ESI+) calcd for C₁₂H₁₅N₂S₂ 251.0677 ([M + H]⁺), found 251.0694.

1-(1-Ethoxyethyl)-1H-benzo[d]imidazole (**3k**) (IIIM/724/1572/CN/ 20).⁷ (Note: reactions with ethers were carried out at room temperature with longer reaction times.) Colorless liquid; TLC R_f = 0.5 (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 8.26 (s, 1H), 7.78 (dd, *J* = 5.3, 2.9 Hz, 1H), 7.60–7.49 (m, 1H), 7.27 (dd, *J* = 5.8, 2.9 Hz, 2H), 5.68 (q, *J* = 5.9 Hz, 1H), 3.45–3.35 (m, 1H), 3.31–3.18 (m, 1H), 1.73 (d, *J* = 6.0 Hz, 3H), 1.10 (t, *J* = 7.0 Hz, 3H); ¹³C NMR (100.61 MHz, CDCl₃) δ 142.8, 140.0, 131.3, 122.2, 121.6, 119.2, 110.0, 82.2, 63.0, 21.1, 13.7; HRMS (ESI+) calcd for C₁₁H₁₅N₂O 191.1184 ([M + H]⁺), found 191.1185.

1-(1-Ethoxyethyl)-5,6-dimethyl-1H-benzo[d]imidazole (**3**I) (IIIM/ 724/1572/CN/66). Colorless liquid; TLC $R_f = 0.5$ (5% MeOH/ DCM); ¹H NMR (400 MHz, CDCl₃) δ 7.83 (s, 1H), 7.49 (s, 1H), 7.27 (s, 1H), 5.52 (q, *J* = 6.0 Hz, 1H), 3.35 (dd, *J* = 15.7, 7.3 Hz, 1H), 3.23 (dd, *J* = 15.1, 7.8 Hz, 1H), 2.30 (d, *J* = 8.1 Hz, 6H), 1.69 (d, *J* = 6.0 Hz, 3H), 1.08 (t, *J* = 6.9 Hz, 3H); ¹³C NMR (100.61 MHz, CDCl₃) δ 142.8, 140.3, 132.2, 131.4, 131.0, 120.3, 111.1, 82.9, 63.9, 22.1, 20.5, 20.2, 14.8; HRMS (ESI+) calcd for C₁₃H₁₉N₂O 219.1497 ([M + H]⁺), found 219.1505.

9-(Tetrahydrofuran-2-yl)-9H-purine (**3r**) (IIIM/724/1572/CN/53). Colorless liquid; TLC $R_f = 0.5$ (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 9.16 (s, 1H), 8.99 (s, 1H), 8.23 (s, 1H), 6.40 (dd, J = 6.2, 2.9 Hz, 1H), 4.32 (dd, J = 14.3, 7.0 Hz, 1H), 4.11 (dd, J = 15.4, 7.6 Hz, 1H), 2.73–2.46 (m, 2H), 2.27–2.11 (m, 2H); ¹³C NMR (100.61 MHz, CDCl₃) δ 152.3, 150.5, 148.5, 143.4, 134.7, 85.9, 69.9, 32.3, 24.3; HRMS (ESI+) calcd for $C_9H_{11}N_4O$ 191.0933 ($[M + H]^+$), found 191.0927.

9-(1,4-Dioxan-2-yl)-9H-purine (**3s**) (IIIM/724/1572/CN/58). Colorless liquid; TLC $R_f = 0.5$ (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 9.11 (s, 1H), 8.95 (s, 1H), 8.46 (s, 1H), 6.05–5.94 (m, 1H), 4.15 (dd, J = 11.8, 1.3 Hz, 1H), 4.06 (dd, J = 12.0, 5.7 Hz, 1H), 3.91–3.81 (m, 4H); ¹³C NMR (100.61 MHz, CDCl₃) δ 152.9, 151.0, 148.8, 143.7, 133.8, 77.2, 68.3, 66.2, 64.3; HRMS (ESI+) calcd for $C_9H_{11}N_4O_2$ 207.0882 ([M + H]⁺), found 207.0878.

9-(Tetrahydrothiophen-2-yl)-9H-purine (**3t**) (IIIM/724/1572/CN/ 59). Colorless liquid; TLC $R_f = 0.5$ (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 9.08 (s, 1H), 8.93 (s, 1H), 8.45 (s, 1H), 6.37– 6.21 (m, 1H), 3.31–3.26 (m, 1H), 3.03–2.97 (m, 1H), 2.41–2.32 (m, 2H), 2.31–2.15 (m, 2H); ¹³C NMR (100.61 MHz, CDCl₃) δ 152.4, 150.9, 148.7, 144.5, 134.9, 62.0, 38.9, 33.3, 28.7; HRMS (ESI+) calcd for C₉H₁₁N₄S 207.0704 ([M + H]⁺), found 207.0696.

9-(1-Ethoxyethyl)-9H-purine (**3u**) (IIIM/724/1572/CN/62). Colorless liquid; TLC $R_f = 0.5$ (5% MeOH/DCM); ¹H NMR (400 MHz, CDCl₃) δ 9.10 (s, 1H), 8.92 (s, 1H), 8.26 (s, 1H), 6.00 (q, J = 6.0 Hz, 1H), 3.56–3.39 (m, 1H), 3.35–3.19 (m, 1H), 1.74 (d, J = 6.0 Hz, 3H), 1.11 (t, J = 7.0 Hz, 3H); ¹³C NMR (100.61 MHz, CDCl₃) δ 152.7, 151.1, 148.7, 142.7, 134.0, 80.8, 64.8, 22.4, 14.7; HRMS (ESI+) calcd for C₉H₁₃N₄O 193.1089 ([M + H]⁺), found 193.1088.

1-(Tetrahydrofuran-2-yl)-1H-benzo[d][1,2,3]triazole (5a) (IIIM/ 724/1572/CN/21(a)).¹⁰ Colorless liquid; TLC $R_f = 0.4$ (30% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.99 (d, J = 8.3Hz, 1H), 7.64 (d, J = 8.3 Hz, 1H), 7.43 (t, J = 7.6 Hz, 1H), 7.31 (t, J = 7.6 Hz, 1H), 6.44 (dd, J = 6.5, 1.6 Hz, 1H), 4.10–3.88 (m, 2H), 3.20– 3.10 (m, 1H), 2.57–2.47 (m, 1H), 2.32–2.43 (m, 1H), 2.19–2.04 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 146.2, 132.8, 127.5, 124.1, 119.7, 110.4, 87.9, 69.2, 30.8, 24.3; HRMS (ESI+) calcd for $C_{10}H_{12}N_3O$ 190.0980 ([M + H]⁺), found 190.0856; MALDI TOF MS calcd for $C_{10}H_{12}N_3O$ ([M + H]⁺) 190.0980, found 190.0971.

2-(Tetrahydrofuran-2-yl)-2H-benzo[d][1,2,3]triazole (**5a**') (IIIM/ 724/1572/CN/21(b)). Colorless liquid; TLC $R_{\rm f}$ = 0.8 (30% EtOAC/ hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.88 (dd, J = 6.4, 2.9 Hz, 2H), 7.38 (dd, J = 6.5, 2.8 Hz, 2H), 6.60 (d, J = 4.8 Hz, 1H), 4.33 (dd, J = 14.0, 7.4 Hz, 1H), 4.13 (dd, J = 14.3, 7.4 Hz, 1H), 2.77–2.54 (m, 1H), 2.54–2.45 (m, 2H), 2.17–2.09 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 144.2, 126.6, 118.4, 94.1, 70.2, 32.3, 24.3; HRMS (ESI +) calcd for C₁₀H₁₂N₃O 190.0980 ([M + H]⁺), found 190.0854; MALDI TOF MS calcd for C₁₀H₁₂N₃O ([M + H]⁺) 190.0980, found 190.0997.

1-(1,4-Dioxan-2-yl)-1H-benzo[d][1,2,3]triazole (**5b**) (IIIM/724/ 1572/CN/29(a)).¹⁰ White solid; mp 85 °C; TLC $R_f = 0.4$ (30% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 8.09 (d, J = 8.3 Hz, 1H), 7.76 (d, J = 8.3 Hz, 1H), 7.53 (t, J = 7.6 Hz, 1H), 7.40 (t, J = 7.6 Hz, 1H), 6.10 (dd, J = 7.2, 2.8 Hz, 1H), 4.54 (dd, J = 11.8, 7.3 Hz, 1H), 4.24 (dd, J = 11.9, 2.7 Hz, 1H), 4.03–3.80 (m, 4H); ¹³C NMR (100.16 MHz, CDCl₃) δ 146.0, 132.5, 127.9, 124.4, 120.1, 110.6, 82.0, 67.3, 65.9, 65.3; HRMS (ESI+) calcd for $C_{10}H_{12}N_3O_2$ 206.0930 ([M + H]⁺), found 206.0925.

2-(1,4-Dioxan-2-yl)-2H-benzo[d][1,2,3]triazole (**5b**') (IIIM/724/1572/CN/29(b)). White solid; mp 85 °C; TLC $R_{\rm f}$ = 0.8 (30% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.91 (dd, *J* = 6.6, 3.0 Hz, 2H), 7.42 (dd, *J* = 6.6, 3.0 Hz, 2H), 6.12 (dd, *J* = 6.4, 2.8 Hz, 1H), 4.52 (dd, *J* = 11.9, 6.5 Hz, 1H), 4.22 (dd, *J* = 11.9, 2.7 Hz, 1H), 4.15–4.07 (m, 1H), 4.05–3.94 (m, 1H), 3.90 (t, *J* = 3.0 Hz, 2H); ¹³C NMR (100.61 MHz, CDCl₃) δ 144.3, 127.1, 118.6, 86.4, 67.6, 65.7, 65.2; HRMS (ESI+) calcd for C₁₀H₁₂N₃O₂ 206.0930 ([M + H]⁺), found 206.0932.

1-(1-Ethoxyethyl)-1H-benzo[d][1,2,3]triazole (5c) (IIIM/724/1572/ CN/22(a)). Colorless liquid; TLC $R_f = 0.5$ (30% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 8.00 (d, J = 8.3 Hz, 1H), 7.73 (d, J = 8.3 Hz, 1H), 7.41 (t, J = 7.6 Hz, 1H), 7.31 (t, J = 7.6 Hz, 1H), 6.19 (q, J = 6.1 Hz, 1H), 3.55–3.40 (m, 1H), 3.24–3.10 (m, 1H), 1.79 (d, J = 6.1 Hz, 3H), 1.06 (t, J = 7.0 Hz, 3H); ¹³C NMR (100.61 MHz, CDCl₃) δ 146.8, 131.1, 127.4, 124.1, 120.0, 111.1, 87.0, 64.3, 21.1, 14.7; HRMS (ESI+) calcd for C₁₀H₁₄N₃O 192.1137 ([M + H]⁺), found 192.1009; GC-MS(EI) *m/z* (relative intensity) 192.2 (M⁺, 2.1), 176.3 (4.65), 147.4 (100), 119.5 (53.98), 91.3 (88.3), 64.4 (37.04), 45.4 (77.61).

2-(1-Ethoxyethyl)-2H-benzo[d][1,2,3]triazole (5*c*') (IIIM/724/ 1572/CN/22(b)). Colorless liquid; TLC $R_f = 0.8$ (30% EtOAC/ hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.83 (dd, J = 5.9, 2.4 Hz, 2H), 7.33 (dd, J = 6.0, 2.3 Hz, 2H), 6.00 (q, J = 5.8 Hz, 1H), 3.59– 3.42 (m, 1H), 3.36–3.17 (m, 1H), 1.85 (d, J = 5.9 Hz, 3H), 1.09 (t, J =7.0 Hz, 3H); ¹³C NMR (100.61 MHz, CDCl₃) δ 144.1, 126.7, 118.5, 91.5, 65.1, 21.5, 14.6; HRMS (ESI+) calcd for C₁₀H₁₄N₃O 192.1137 ([M + H]⁺), found 192.1009; GC–MS(EI) *m*/*z* (relative intensity) 192.2 (M⁺, 2.2), 176.3 (5.7), 147.4 (100), 119.4 (68.2), 91.3 (86.6), 77.3 (33.54), 64.4 (41).

1-(Tetrahydrothiophen-2-yl)-1H-benzo[d][1,2,3]triazole (5d) (IIIM/724/1572/CN/24). Colorless liquid; TLC $R_f = 0.5$ (30% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.99 (d, J = 8.3Hz, 1H), 7.59 (d, J = 8.3 Hz, 1H), 7.42 (t, J = 7.6 Hz, 1H), 7.31 (t, J = 7.6 Hz, 1H), 6.44–6.35 (m, 1H), 3.31–3.20 (m, 1H), 3.13–2.99 (m, 1H), 2.94–2.84 (m, 1H), 2.57–2.36 (m, 2H), 2.32–2.22 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 146.4, 132.1, 127.1, 123.9, 120.3, 110.3, 65.9, 37.1, 33.8, 29.9; HRMS (ESI+) calcd for C₁₀H₁₂N₃S 206.0752 ([M + H]⁺), found 206.0621; MALDI TOF MS calcd for C₁₀H₁₁N₃S ([M + H]⁺) 206.0752, found 206.0724.

1-(Tetrahydro-2H-pyran-2-yl)-1H-benzo[d][1,2,3]triazole (**5f**) (IIIM/724/1582/CN/01). Colorless liquid; TLC $R_f = 0.4$ (30% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 8.08 (d, J = 8.4Hz, 1H), 7.75 (d, J = 8.3 Hz, 1H), 7.50 (t, J = 7.6 Hz, 1H), 7.39 (t, J = 7.6 Hz, 1H), 6.05 (dd, J = 8.2, 2.8 Hz, 1H), 3.96 (dd, J = 11.6, 4.4 Hz, 1H), 3.86–3.72 (m, 1H), 2.71–2.56 (m, 1H), 2.29–2.11 (m, 2H), 1.87–1.68 (m, 3H); ¹³C NMR (100.61 MHz, CDCl₃) δ 146.2, 132.5, 127.4, 124.1, 119.8, 111.1, 85.6, 66.8, 29.2, 24.9, 21.4; HRMS (ESI+) calcd for C₁₁H₁₃N₃O 204.1137 ([M + H]⁺), found 204.1002; MALDI TOF MS calcd for C₁₁H₁₃N₃O ([M + H]⁺) 204.1137, found 204.1145.

1-(*Tetrahydrofuran*-2-yl)-1H-1,2,3-triazole (**5g**) (IIIM/724/1572/ CN/30(a)). Colorless liquid; TLC $R_f = 0.3$ (30% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.71 (s, 2H), 6.25 (d, J = 6.5 Hz, 1H), 4.18–3.99 (m, 2H), 2.80–2.65 (m, 1H), 2.50–2.40 (m, 1H), 2.16– 2.09 (m, 2H); ¹³C NMR (100.61 MHz, CDCl₃) δ 133.6, 122.0, 89.4, 69.6, 32.2, 23.8; HRMS (ESI+) calcd for $C_6H_{10}N_3O$ 140.0824 ([M + H]⁺), found 140.0771.

2-(*Tetrahydrofuran-2-yl*)-2H-1,2,3-triazole (**5g**') (*IIIM*/724/1572/ *CN*/30(b)). Colorless liquid; TLC $R_f = 0.7$ (30% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.63 (s, 2H), 6.39–6.26 (m, 1H), 4.16 (dd, J = 13.9, 7.4 Hz, 1H), 4.04 (dd, J = 13.9, 7.6 Hz, 1H), 2.72–2.56 (m, 1H), 2.45–2.36 (m, 2H), 2.20–2.01 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 134.5, 92.1, 69.5, 31.3, 24.4; HRMS (ESI+) calcd for C₆H₁₀N₃O 140.0824 ([M + H]⁺), found 140.0815; MALDI TOF MS calcd for C₆H₉N₃O ([M + H]⁺) 140.0824, found 140.0839.

1-(1,4-Dioxan-2-yl)-1H-1,2,3-triazole (5h) (IIIM/724/1572/CN/ 31(a)). Colorless liquid; TLC $R_f = 0.3$ (30% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.85 (s, 1H), 7.70 (s, 1H), 5.84 (s, 1H), 4.12 (dd, J = 11.9, 2.0 Hz, 1H), 4.01 (dd, J = 11.9, 5.9 Hz, 1H), 3.91– 3.67 (m, 4H); ¹³C NMR (100.61 MHz, CDCl₃) δ 133.7, 122.7, 81.8, 68.2, 66.0, 64.3; HRMS (ESI+) calcd for C₆H₁₀N₃O₂ 156.0773 ([M + H]⁺), found 156.0764.

2-(1,4-Dioxan-2-yl)-2H-1,2,3-triazole (5h') (IIIM/724/1572/CN/ 31(b)). Colorless liquid; TLC $R_f = 0.7$ (30% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.65 (s, 2H), 5.78 (dd, J = 7.1, 2.6 Hz, 1H), 4.27 (dd, J = 11.7, 7.3 Hz, 1H), 4.03 (dd, J = 11.8, 2.5 Hz, 1H), 3.95–3.84 (m, 2H), 3.80–3.70 (m, 2H); ¹³C NMR (100.61 MHz, CDCl₃) δ 135.2, 84.8, 67.5, 65.8, 65.4; HRMS (ESI+) calcd for C₆H₁₀N₃O₂ 156.0773 ([M + H]⁺), found 156.0764.

4-Phenyl-1-(tetrahydrofuran-2-yl)-1H-1,2,3-triazole (5i) (IIIM/ 724/1572/CN/39(a)). Colorless liquid; TLC $R_f = 0.3$ (30% EtOAC/ hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.77 (s, 1H), 7.70 (d, J = 7.5Hz, 2H), 7.31 (t, J = 7.4 Hz, 2H), 7.24 (t, J = 7.2 Hz, 1H), 6.26–6.19 (m, 1H), 4.09 (dd, J = 14.3, 7.4 Hz, 1H), 3.94 (dd, J = 14.1, 7.4 Hz, 1H), 2.63–2.52 (m, 1H), 2.41–2.22 (m, 2H), 2.03–1.88 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 148.0, 131.4, 130.3, 128.8, 128.5, 126.0, 92.3, 69.5, 31.3, 24.5; HRMS (ESI+) calcd for C₁₂H₁₄N₃O 216.1137 ([M + H]⁺), found 216.1134. 4-Phenyl-2-(tetrahydrofuran-2-yl)-2H-1,2,3-triazole (5i') (IIIM/ 724/1572/CN/39(b)). Colorless liquid; TLC $R_f = 0.7$ (30% EtOAC/ hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.79 (s, 1H), 7.75–7.69 (m, 2H), 7.35 (t, J = 7.3 Hz, 2H), 7.27 (t, J = 7.3 Hz, 1H), 6.29–6.23 (m, 1H), 4.13 (dd, J = 14.4, 7.3 Hz, 1H), 3.99 (dd, J = 14.1, 7.3 Hz, 1H), 2.68–2.56 (m, 1H), 2.45–2.27 (m, 2H), 2.10–1.96 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 148.0, 131.4, 130.3, 128.8, 128.5, 126.0, 92.3, 69.5, 31.3, 24.4; HRMS (ESI+) calcd for C₁₂H₁₄N₃O 216.1137 ([M + H]⁺), found 216.1132; MALDI TOF MS calcd for C₁₂H₁₃N₃O ([M + H]⁺) 216.1137, found 216.1156.

1-(*Tetrahydrofuran*-2-*yl*)-1*H*-1,2,4-triazole (**5***j*) (*IIIM*/724/1572/CN/34).¹ Colorless liquid; TLC $R_{\rm f}$ = 0.4 (50% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 8.28 (s, 1H), 7.97 (s, 1H), 6.12–6.02 (m, 1H), 4.18 (dd, *J* = 13.4, 8.0 Hz, 1H), 4.04 (dd, *J* = 15.4, 7.6 Hz, 1H), 2.60–2.53 (m, 1H), 2.43–2.33 (m, 1H), 2.21–1.99 (m, 2H); ¹³C NMR (100.61 MHz, CDCl₃) δ 151.7, 142.1, 88.9, 69.6, 32.2, 23.5; HRMS (ESI+) calcd for $C_6H_{10}N_3O$ 140.0824 ([M + H]⁺), found 140.0829.

1-(*Tetrahydrofuran-2-yl*)-1*H*-pyrazole (5k) (IIIM/724/1572/CN/ 47). Colorless liquid; TLC $R_{\rm f}$ = 0.4 (20% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.48 (s, 2H), 6.20 (s, 1H), 5.93 (d, *J* = 6.3 Hz, 1H), 4.05 (dd, *J* = 14.1, 7.6 Hz, 1H), 3.91 (dd, *J* = 14.8, 7.4 Hz, 1H), 2.58–2.45 (m, 1H), 2.34–2.22 (m, 1H), 2.18–2.04 (m, 1H), 2.01–1.93 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 139.8, 128.0, 105.6, 90.0, 69.1, 31.7, 24.4; HRMS (ESI+) calcd for C₇H₁₀N₂O 139.0871 ([M + H]⁺), found 139.0859.

3-Methyl-1-(tetrahydrofuran-2-yl)-1H-pyrazole (5I) (IIIM/724/ 1572/CN/50). Colorless liquid; TLC R_f = 0.4 (20% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.37 (s, 1H), 5.97 (s, 1H), 5.89–5.79 (m, 1H), 4.05 (dd, *J* = 14.3, 7.0 Hz, 1H), 3.89 (dd, *J* = 14.5, 7.2 Hz, 1H), 2.56–2.41 (m, 1H), 2.34–2.16 (m, 4H), 2.14–2.03 (m, 1H), 2.01–1.91 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 149.2, 128.6, 105.4, 89.8, 69.0, 31.6, 24.5, 13.5; HRMS (ESI+) calcd for C₈H₁₃N₂O 153.1028 ([M + H]⁺), found 153.1025.

3-Phenyl-1-(tetrahydrofuran-2-yl)-1H-pyrazole (5m) (IIIM/724/ 1572/CN/44). Colorless liquid; TLC $R_f = 0.4$ (20% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.74 (d, J = 7.4 Hz, 2H), 7.51 (d, J =1.9 Hz, 1H), 7.31 (t, J = 7.5 Hz, 2H), 7.21 (dd, J = 14.3, 6.8 Hz, 1H), 6.50 (d, J = 2.0 Hz, 1H), 5.96 (dd, J = 6.3, 1.8 Hz, 1H), 4.11 (dd, J =13.6, 7.8 Hz, 1H), 3.94 (dd, J = 15.0, 7.4 Hz, 1H), 2.65–2.51 (m, 1H), 2.33–2.23 (m, 1H), 2.21–2.09 (m, 1H), 2.02–1.94 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 151.7, 133.6, 129.1, 128.5, 127.4, 125.7, 102.9, 90.3, 69.2, 31.9, 24.3; HRMS (ESI+) calcd for C₁₃H₁₅N₂O 215.1184 ([M + H]⁺), found 215.1198.

4-(4-Bromophenyl)-1-(tetrahydrofuran-2-yl)-1H-pyrazole (**5n**) (*IIIM*/724/1572/CN/37). Colorless liquid; TLC $R_f = 0.4$ (20% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 7.72 (d, J = 4.1 Hz, 2H), 7.61 (s, 1H), 7.53 (d, J = 7.0 Hz, 2H), 6.57 (d, J = 1.3 Hz, 1H), 6.05 (d, J = 5.2 Hz, 1H), 4.20 (dd, J = 12.5, 7.0 Hz, 1H), 4.04 (dd, J = 14.2, 7.0 Hz, 1H), 2.74–2.62 (m, 1H), 2.43–2.34 (m, 1H), 2.32–2.18 (m, 1H), 2.10–2.01 (m, 1H); ¹³C NMR (100.61 MHz, CDCl₃) δ 150.7, 132.6, 131.6, 129.3, 127.2, 121.4, 102.8, 90.3, 69.4, 32.0, 24.3; HRMS (ESI+) calcd for C₁₃H₁₄BrN₂O 293.0290 ([M + H]⁺), found 293.0287.

2,2,6,6-Tetramethyl-1-((tetrahydrofuran-2-yl)oxy)piperidine (6) (IIIM/724/1572/CN/55). Colorless liquid; TLC $R_f = 0.8$ (5% EtOAC/hexane); ¹H NMR (400 MHz, CDCl₃) δ 5.29 (d, J = 4.5 Hz, 1H), 3.84–3.69 (m, 2H), 2.00–1.78 (m, 4H), 1.40 (d, J = 6.5 Hz, 6H), 1.20–0.87 (m, 12H); ¹³C NMR (100.61 MHz, CDCl₃) δ 109.5, 66.6, 60.1, 58.6, 40.0, 39.6, 33.8, 33.3, 31.2, 23.8, 20.4, 20.0, 17.2; HRMS (ESI+) calcd for $C_{13}H_{26}NO_2$ 228.1964 ([M + H]⁺), found 228.1961.

ASSOCIATED CONTENT

S Supporting Information

Copies of NMR and MS spectra, XRD analysis details, and a CIF file for **3i**. This material is available free of charge via the Internet at http://pubs.acs.org.

The Journal of Organic Chemistry

Corresponding Author

*Tel.: +91-191-2569001-010 (292). Fax: +91-191-2569333. Email: ppsingh@iiim.ac.in.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

The authors acknowledge the financial support of CSIR through Research Grants BSC 0108 and HCP 0001. H.A., U.S., S.S. and S.G. thank CSIR, UGC, DBT, and OSDD for their fellowships. This article is IIIM Communication No. IIIM/1729/2014.

REFERENCES

(1) (a) Liao, C.-L.; Shie, J.-J.; Liang, P.-H.; Wong, C.-H. Chem. Biol. 2006, 13, 261-268. (b) Ren, Y.; Zhang, L.; Zhou, C.-H.; Geng, R.-X. Med. Chem. 2014, 4, 640-662. (c) Rad, M. N. S.; Khalafi-Nezhad, A.; Behrouz, S. Beilstein J. Org. Chem. 2010, 6, No. 49. (d) Zhang, H.-Z.; Damu, G. L. V.; Cai, G.-X.; Zhou, C.-H. Eur. J. Med. Chem. 2013, 64, 329-344. (e) Rezaei, Z.; Khabnadideh, S.; Pakshir, K.; Hossaini, Z.; Amiri, F.; Assadpour, E. Eur. J. Med. Chem. 2009, 44, 3064-3067. (f) Sharma, D.; Narasimhan, B.; Kumar, P.; Jalbout, A. Eur. J. Med. Chem. 2009, 44, 1119-1127. (g) Santo, R. D.; Tafi, A.; Costi, R.; Botta, M.; Artico, M.; Corelli, F.; Forte, M.; Caporuscio, F.; Angiolella, L.; Palamara, A. T. J. Med. Chem. 2005, 48, 5140-5153. (h) Güven, Ö. Ö.; Erdoğan, T.; Göker, H.; Yıldız, S. Bioorg. Med. Chem. Lett. 2007, 17, 2233-2236. (i) Rossello, A.; Bertini, S.; Lapucci, A.; Macchia, M.; Martinelli, A.; Rapposelli, S.; Herreros, E.; Macchia, B. J. Med. Chem. 2002, 45, 4903-4912. (j) Dyer, R. L.; Ellames, G. J.; Hamill, B. J.; Manley, P. W.; Pope, A. M. S. J. Med. Chem. 1983, 26, 442-445. (k) Beena; Kumar, N.; Rohilla, R. K.; Roy, N.; Rawat, D. S. Bioorg. Med. Chem. Lett. 2009, 19, 1396-1398. (1) Mizuno, C. S.; Chittiboyina, A. G.; Shah, F. H.; Patny, A.; Kurtz, T. W.; Pershadsingh, H. A.; Speth, R. C.; Karamyan, V. T.; Carvalho, P. B.; Avery, M. A. J. Med. Chem. 2010, 53, 1076-1085. (m) Bovens, S.; Kaptur, M.; Elfringhoff, A. S.; Leh, M. Bioorg. Med. Chem. Lett. 2009, 19, 2107-2111. (n) Dubey, A.; Srivastava, S. K.; Srivastava, S. D. Bioorg. Med. Chem. Lett. 2011, 21, 569-573. (o) Giornal, F.; Pazenok, S.; Rodefeld, L.; Lui, N.; Vors, J.-P.; Leroux, F. R. J. Fluorine Chem. 2013, 152, 2-11.

(2) (a) Li, F.; Hu, J. J.; Koh, L. L.; Hor, T. S. A. Dalton Trans. 2010, 39, 5231–5241. (b) Huckaba, A. J.; Hollis, T. K.; Howell, T. O.; Valle, H. U.; Wu, Y. Organometallics 2013, 32, 63–69. (c) Gupta, S.; Basu, B.; Das, S. Tetrahedron 2013, 69, 122–128. (d) Kore, R.; Srivastava, R. J. Mol. Catal. A: Chem. 2011, 345, 117–126. (e) Tsuji, Y.; Ohno, H. RSC Adv. 2012, 2, 11279–11284.

(3) (a) Milen, M.; Grün, A.; Bálint, E.; Dancsó, A.; Keglevich, G. *Synth. Commun.* **2010**, *40*, 2291–2301. (b) Shieh, W.-C.; Lozanov, M.; Repič, O. *Tetrahedron Lett.* **2003**, *44*, 6943–6945. (c) Hayat, S.; Attaur-Rahman; Choudhary, M. I.; Khan, K. M.; Schumann, W.; Bayer, E. *Tetrahedron* **2001**, *57*, 9951–9957.

(4) (a) Guo, S.; Yu, J.-T.; Dai, Q.; Yang, H.; Cheng, J. Chem. Commun. 2014, 50, 6240–6242. (b) Jia, Z.; Nagano, T.; Li, X.; Chan, A. S. C. Eur. J. Org. Chem. 2013, 858–861. (c) Chu, X.-Q.; Meng, H.; Zi, Y.; Xu, X.-P.; Ji, S.-J. Chem. Commun. 2014, 50, 9718–9721.

(5) (a) Majji, G.; Guin, S.; Gogoi, A.; Rout, S. K.; Patel, B. K. Chem. Commun. 2013, 49, 3031–3033. (b) Shi, E.; Shao, Y.; Chen, S.; Hu, H.; Liu, Z.; Zhang, J.; Wan, X. Org. Lett. 2012, 14, 3384–3387.
(c) Zhang, S.; Guo, L.-N.; Wang, H.; Duan, X.-H. Org. Biomol. Chem. 2013, 11, 4308–4311. (d) Huang, J.; Li, L.-T.; Li, H.-Y.; Husan, E.; Wang, P.; Wang, B. Chem. Commun. 2012, 48, 10204–10206.

(6) (a) Zhang, X.; Wang, M.; Li, P.; Wang, L. Chem. Commun. 2014, 50, 8006–8009. (b) Xue, Q.; Xie, J.; Li, H.; Cheng, Y.; Zhu, C. Chem. Commun. 2013, 49, 3700–3702. (c) Zhao, J.; Li, P.; Xia, C.; Li, F. Chem. Commun. 2014, 50, 4751–4754. (d) Chen, S.; Xu, Y.; Wan, X. Org. Lett. 2011, 13, 6152–6155. (e) Froehr, T.; Sindlinger, C. P.;

Kloeckner, U.; Finkbeiner, P.; Nachtsheim, B. J. Org. Lett. **2011**, *13*, 3754–3757. (f) Mai, W.-P.; Song, G.; Yuan, J.-W.; Yang, L.-R.; Sun, G.-C.; Xiao, Y.-M.; Mao, P.; Qu, L.-B. RSC Adv. **2013**, *3*, 3869–3872. (g) Lv, Y.; Li, Y.; Xiong, T.; Lu, Y.; Liu, Q.; Zhang, Q. Chem. Commun. **2014**, 50, 2367–2369.

(7) Pan, S.; Liu, J.; Li, H.; Wang, Z.; Guo, X.; Li, Z. Org. Lett. 2010, 12, 1932–1935.

(8) Wu, X.-F.; Gong, J.-L.; Qi, X. Org. Biomol. Chem. 2014, 12, 5807–5817.

(9) Singh, P. P.; Aithagani, S. K.; Yadav, M.; Singh, V. P.; Vishwakarma, R. A. J. Org. Chem. 2013, 78, 2639–2648.

(10) Xiang, L.; Yongxin, C.; Kangning, L.; Dong, W.; Baohua, C. Chin. J. Chem. 2012, 30, 2285.